

# Tutorial Energy & Sustainability

## Exercise Sheet 4 –Sketch of Solution–

### Exercise 1

$$\text{Global: } \frac{184\,246 \text{ Mt} + 991\,241 \text{ Mt}}{911.5 \text{ Mt/a}} = 1\,289.6 \text{ years}$$

$$\text{D: } \frac{41\,300 \text{ Mt} + 77\,600 \text{ Mt}}{179.1 \text{ Mt/a}} = 663.87 \text{ years}$$

### Exercise 2

$$1.67 \cdot 10^{12} \text{ kg HCU} \cdot \frac{1}{1022 \frac{\text{kg HCU}}{\text{t hard coal}}} = 1630 \cdot 10^6 \text{ t of hard coal}$$

$$728\,436 \cdot 10^6 \text{ t of hard coal} = 1630 \cdot 10^6 \text{ t of hard coal} \cdot \sum_{t=0}^m 1.03^t$$

$$446 = \frac{1.03^{n+1} - 1}{1.03 - 1}$$

$$13.4 = 1.03^{n+1} - 1$$

$$14.4 = 1.03^{n+1}$$

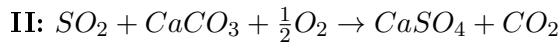
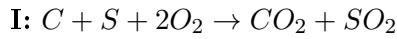
$$n + 1 = \log_{1.03} 14.4 = 90 \Rightarrow n = 89 \text{ years}$$

### Exercise 3

cf. tutorial

### Exercise 4

a) (strongly simplified)



[ $CaCO_3$  – calcium carbonate;  $CaSO_4$  – calcium sulfate/cement]

b) 25 t sulfur  $\rightarrow$  Change units to mol!

**III:** amount of substance  $n = \frac{\text{mass m}}{\text{molar mass M}}$

$$\frac{25 \cdot 10^6 \text{ g}}{32 \text{ g/mol}} S = 781\,250 \text{ mol} = n(S)$$

With I and II the following equation olds:

$$n(S) = n(SO_2) \text{ and } n(SO_2) = n(CaCO_3)$$

From III follows  $m(CaCO_3) = n(CaCO_3) \cdot M(CaCO_3) = 78 \cdot 10^6 \text{ g}$

$$\approx 78 \text{ t } CaCO_3$$

c)  $m(CO_2) = n(CO_2) \cdot M(CO_2) = 34 \text{ t}$   
(with  $n(CO_2) = n(S)$  and  $M(CO_2) = 44 \text{ g/mol}$ )

## Exercise 5

- a) Depletion in D =  $63 \text{ Mt} \cdot 0.37 = 23.31 \text{ Mt}$   
prices on the world market =  $65 \text{ USD/t} \cdot \frac{1 \text{ Euro}}{1.2 \text{ USD}} = 54.17 \text{ Euro/t}$   
Difference to the german prices =  $85.83 \text{ Euro/t}$   
Subsidies =  $85.83 \text{ Euro/t} \cdot 23.31 \text{ Mt} = 2 \cdot 10^9 \text{ Euro p.a.}$
- b) approx. 80 000 Euro per employee and year.